

# Sedimentary environmental quality of a biosphere reserve estuary from Southwestern Iberian Peninsula

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## ABSTRACT

The Huelva estuary is formed by the common mouths of the Odiel and Tinto Rivers, and inside this ecosystem is the biosphere reserve of the Odiel saltmarshes. This ecosystem has been historically affected by acid mine drainage (AMD) and by releases of pollutants from five phosphoric acid industrial plants and phosphogypsum (PG) waste stacks located in the area. Therefore, this study aims to carry out a comprehensive assessment of the environmental impact of the biosphere reserve of the Odiel saltmarshes. For this, it was necessary to find a suitable sedimentary background (Piedras River in our case). To quantify this impact, several pollution indexes were used. According to the values reached by the indexes, this impact was classified as “serious” pollution for most trace elements, excepting the deepest layers, and “low-moderate” pollution for the <sup>238</sup>U-series radionuclides, while no pollution for the <sup>232</sup>Th-series and <sup>40</sup>K radionuclides was found as expected.

**Keywords:** Phosphogypsum piles; Fertilizer industry; Acid mine drainage; Trace elements; Natural radionuclides; Pollution indexes

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## 1. Introduction

The estuary of Huelva is in the Southwestern Spain, and it is formed by the common mouths of the Tinto and Odiel Rivers (see Fig. 1), which are ones of the most polluted rivers in the world by acid mine drainage (AMD). Their waters contain very high levels of concentrations (3-5 orders of magnitude higher than unperturbed surface waters) of heavy metals and natural radionuclides, especially for U and Th isotopes (Guerrero et al., 2021a; Olías et al., 2006; Nieto et al., 2013; Nieto et al., 2007). In addition, in this estuary a very large industrial chemical complex is also located, whose activity began in 1965, including plants devoted to the production of fertilizers, ammonia, copper by smelting/electrorefining, oil refining, petrochemical, TiO<sub>2</sub> pigments, etc. Both pollution sources have produced a high historical pollution of this estuary from the XIX century to the present day (Borrego et al., 2002; Morillo et al., 2004; Pérez-López et al., 2011).

40 A big fraction of the basins of these rivers is in the Iberian pyrite belt (IPB), southwest of the  
41 Iberian Peninsula, one of the major deposits of massive polymetallic sulphurs in the world, being  
42 estimated a polymetallic sulphur reserve of about 1.7 Gt (Sáez et al., 1999). These deposits were  
43 mined by other civilizations such as Tartessos and Roman, but the greatest mining activity started  
44 in the second half of XIX century, mainly by British and French mining companies, generating in  
45 the North of the Huelva province old mining galleries, millings, and several types of waste  
46 (Mujica et al., 2008). These “legacy sites” contain the sources of the AMD problem in both Odiel  
47 and Tinto Rivers waters because the basins of these rivers receive waters from the IPB.

48 AMD is produced by the oxidation of these residual materials that interact with the water and its  
49 dissolved oxygen, generating the release of  $\text{Fe}^{2+}$  and a great acidification of the aqueous medium  
50 reaching pH up to 1-2 (Aduvire, 2006). These sulphurs are mainly composed by iron pyrite ( $\text{FeS}_2$ ),  
51 and variable proportions of other subordinate minerals as sphalerite (ZnS), galena (PbS),  
52 chalcopyrite ( $\text{CuFeS}_2$ ), arsenopyrite ( $\text{FeAsS}$ ), pyrrhotite ( $\text{Fe}_{1-x}\text{S}$ ), and many other minor phases,  
53 such as Bi- and Pb-sulfosalts, cassiterite, magnetite, stannite, electrum and cobaltite. The AMD  
54 process makes those minor elements contained in the sulphur minerals and their host-rocks to be  
55 released into the aqueous matrix (heavy metals, rare earth elements, radionuclides, etc.),  
56 generating very polluted waters (Curcio et al., 2019).

57 Another pollution problem in the Huelva estuary is related to the chemical industrial complex  
58 located in its vicinity, whose plants have released huge amounts of pollutants, mainly heavy  
59 metals, natural radionuclides, and anions into the Huelva estuary. In relation to the phosphoric  
60 acid production, 5 production plants were installed in the estuarine zone, using phosphate rock  
61 (PR) from Morocco as main raw material. The Moroccan phosphate rock is characterized by  
62 having a high concentration of  $^{238}\text{U}$  (around  $1500 \text{ Bq kg}^{-1}$ , equivalent to about 50 ppm of natural  
63 U), which is about 50 times higher than the concentration of  $^{238}\text{U}$  for a non-polluted soil  
64 (UNSCEAR, 2000). During the process, a by-product called phosphogypsum (PG) is generated,  
65 containing more than 95% of the majority of radioactivity initially contained in the raw material  
66 ( $^{226}\text{Ra}$ ,  $^{230}\text{Th}$  and  $^{210}\text{Pb}$ ), but only about 10-20% of  $^{238}\text{U}$  (Bolívar et al., 2009; Gázquez et al., 2009).

67 Three different periods are distinguished in the PG management. The first one goes from 1965 to  
68 1997, when the 20% of the PG was discharged directly into the Odiel channel and the remaining  
69 80% was transported by using estuarine water, and finally deposited in piles in the right shore of  
70 the Tinto River channel, while the water used in the transport, with a pH around 1.5 and high  
71 concentration of metals and radionuclides, was returned into the Tinto River (Bolívar et al., 2008;  
72 Bolívar et al., 1995; García-Tenorio and García-León, 1996; Gázquez et al., 2009). During the  
73 second period (1998 to the end of 2010), the full PG generated in the 5 plants of phosphoric acid  
74 was pumped by using freshwater in a closed circuit since the used water was pumped back into  
75 the factories to be employed newly for PG pumping. And finally, on December 31<sup>st</sup>, 2010, the PA  
76 production was stopped, and since this date there has been no generation of phosphogypsum in  
77 Huelva (Hierro et al., 2012).

78 The main problem related to the PG direct discharges into the Odiel channel, and to a lesser extent  
79 the leachates from PG waste piles, are the releases of a great number of natural radionuclides such  
80 as  $^{238}\text{U}$ ,  $^{226}\text{Ra}$ ,  $^{210}\text{Pb}$  or  $^{210}\text{Po}$ , and toxic elements like Fe, Zn, As, Cr or Cd to their surroundings  
81 (Papaslioti et al., 2018; Pérez-López et al., 2016; Guerrero et al., 2021a,b)). Nowadays there are  
82 around 100 Mt of phosphogypsum distributed along 1000 ha on the occidental shore of the Tinto  
83 River, divided into 4 different areas (Hierro et al., 2013).

84 For these reasons, that is, due to both AMD releases and phosphogypsum leachates (PGL), the  
85 estuary of Huelva has a significant pollution level, which is caused by releases of trace elements  
86 and natural radionuclides. This can be corroborated by other studies on the pollution in the Odiel  
87 and Tinto Rivers (Borrego et al., 2002; Pérez-López et al., 2011).

88 Consequently, the aim of this study is to assess and update the environmental impact on the Odiel  
89 River estuarine biosphere reserve and for that, vertical profiles were analyzed. Furthermore,  
90 different pollution indexes were obtained in order to assess the environmental impact, which was  
91 possible due to the proper selection of a sedimentary background, also called baseline, that needs  
92 to be geochemically similar to the problem sedimentary system (Le Gall et al., 2018; Luo et al.,  
93 2022).

94 Considering everything previously mentioned, to the best of our knowledge, this is the most  
95 updated study on the environmental impact of the estuarine biosphere reserve from the  
96 Southwestern Iberian Peninsula. In addition, an exhaustive assessment and establishment of a  
97 proper sedimentary background (Piedras River in our case) was carried out, and the comparison  
98 with several sedimentary backgrounds placed at the Southwestern Iberian Peninsula was made.  
99 Moreover, the quantification of the impact in Bacuta Island was carefully accomplished using  
100 several pollution and toxicity indexes for major and trace elements, heavy metals and natural  
101 radionuclides, where the different pollution sources were identified by using correlation analysis.  
102 Furthermore, the procedure followed in this study to select a suitable sedimentary background,  
103 assess the environmental impact in the sedimentary system of interest and identify the different  
104 pollution sources can be applied to any sedimentary system worldwide, which makes this study  
105 be even more applicable.

106

## 107 **2. Materials and methods**

### 108 **2.1 Sampling**

109 Samplings were carried out in November 2021, taking a sediment core from Bacuta Island in the  
110 biosphere reserve of the Odiel saltmarshes and another one from the Piedras River, using this  
111 latter as sedimentary reference system to assess the background of this geographical area. This  
112 was necessary because it must have similar geochemical substrate that Huelva estuary, and do not  
113 contain a detectable influence from AMD or PG stacks and chemical complex (see Fig. 1). Every  
114 core had about 60 cm of depth and was cut into 2 cm thick sections (Fig. A.1, in Supplementary  
115 Material – Appendix A). Then, the 2 cm thick samples were dried in an oven at 60 °C until constant  
116 weight, grinded by using an agate mortar, and finally well homogenized.



**Fig. 1.** Study area selected for the samplings carried out in this work, which the sediment cores were taken from Piedras River and Bacuta Island.

## 117 2.2 Characterization techniques

118 The major elements were analyzed by X-ray fluorescence (XRF) using the instrument Panalytical  
 119 Spectrometer AXIOS at the Research, Technology and Innovation Centre of the University of  
 120 Seville (CITIUS). Trace elements were analysed by inductively coupled plasma mass  
 121 spectrometry (ICP-MS) at Activation Laboratories (Actlabs, Ontario, Canada), for which the  
 122 samples were previously digested with strong acids such as hydrofluoric acid, followed by a  
 123 mixture of nitric and perchloric acids. Granulometric analysis was carried out by laser diffraction,  
 124 using the equipment Malvern Mastersizer 2000 by the General Services of the University of  
 125 Huelva. The radionuclides by gamma-ray spectrometry were determined by using a well-type  
 126 high purity germanium (HPGe) detector and an extended range HPGe detector (XtRa), whose  
 127 efficiency calibrations were developed by Barba-Lobo et al. (2021a) and Barba-Lobo et al.

128 (2021b), respectively. This technique is widely used by the Radiation Physics and Environment  
129 Research Group (FRYMA) of the University of Huelva.

### 130 **2.3 Quality control**

131 All the measurements that were not carried out by the research group FRYMA, were ordered to  
132 accredited external laboratories. For every measurement, one replica and one blank per each ten  
133 samples were included, as well as certificated reference materials (CRMs) provided by the IAEA  
134 (International Atomic Energy Agency), where two CRMs were of soil type (IAEA-326 and IAEA-  
135 327) and another of phosphogypsum type (IAEA-434), obtaining generally  $z_{score}$  values less than  
136 2. In addition, the laboratory of the group FRYMA annually participates in several  
137 intercomparison exercises, where gamma-ray spectrometry was employed for the analysis. To  
138 guarantee the reliability of the results, many different comparisons between different analysis  
139 techniques were accomplished such as the comparison between ICP-MS and gamma-ray  
140 spectrometry. Regarding the detection limits obtained for ICP-MS and gamma-ray spectrometry,  
141 they were ranged from 0.05 ppm and 0.1 ppm, and from 2 Bq kg<sup>-1</sup> to 20 Bq kg<sup>-1</sup>, respectively.

142 In addition, the validation for gamma-ray spectrometry was also accomplished by comparing the  
143 <sup>40</sup>K concentration obtained by using gamma-ray spectrometry with that concentration obtained by  
144 means of XRF using the relationship: 1% of natural K → 313 mBq/g of <sup>40</sup>K (Barba-Lobo et al.,  
145 2021a).

### 146 **2.4 Pollution and toxicity indexes**

147 To evaluate the environmental impact in the Huelva estuarine by AMD and phosphogypsum  
148 discharges, several pollution indexes have been used and the classification for every pollution  
149 index is shown in Table A.1 (Supplementary Material – Appendix A).

#### 150 **2.4.1 Enrichment factor (EF)**

151 The enrichment factor for an element (i) in a sample (s),  $EF_i$ , has been used to evaluate the degree  
152 of anthropic pollution in soils and sediments by using a conservative element as normalizer of the  
153 concentrations, being Al the most one used (Díaz-Asencio et al., 2011; Hakason, 1979; Lario et  
154 al., 2016; Zhang et al., 2016). The  $EF$  can be calculated by Eq. 1:

$$EF_i = \frac{(C_i/C_{Al})_s}{(C_i/C_{Al})_r} \quad (1)$$

155

156 where  $C_i$  is the concentration of a specific element (i) in the sample (s),  $(C_i)_s$ , and in the reference  
157 (r),  $(C_i)_r$ , and  $C_{Al}$  is the concentration of Al in the sample,  $(C_{Al})_s$ , and in the reference,  $(C_{Al})_r$ .

#### 158 **2.4.2 Contamination factor (CF)**

159 To estimate the pollution level about a specific element with respect a typical soil, the  
160 contamination factor,  $CF$ , is calculated using the following equation (Gözel et al., 2022;  
161 Vineethkumar et al., 2020):

$$CF_i = \frac{(C_i)_s}{(C_i)_r} \quad (2)$$

162

163 where  $(C_i)_s$  is the concentration of an element (i) present in the sample (s) and  $(C_i)_r$  is referred to  
164 the concentration of this element reference (r) system, which in our case was taken from (Rudnick  
165 and Gao, 2003).

166 For the *CF* assessment in the case of natural radionuclides, that is,  $^{238}\text{U}$ -series,  $^{232}\text{Th}$ -series and  
167  $^{40}\text{K}$ , the activity concentrations established by UNSCEAR (2000) for unpolluted typical soils, that  
168 is,  $35 \text{ Bq kg}^{-1}$ ,  $30 \text{ Bq kg}^{-1}$  and  $400 \text{ Bq kg}^{-1}$  for  $^{238}\text{U}$ ,  $^{232}\text{Th}$  and  $^{40}\text{K}$ , respectively, were used as  
169 reference activity concentrations.

### 170 **2.4.3 Contamination degree (CD)**

171 The contamination degree (*CD*) is a factor to measure the mean level of contamination of a  
172 sample/material by considering all the toxic elements, such as As, Cd, Cr, Cu, Ni, Pb and Zn, and  
173 can be calculated by the following equation (Vineethkumar et al., 2020; Yushin et al., 2023):

$$CD = \frac{\sum_{i=1}^n FC_i}{n} \quad (3)$$

174 where  $n$  is the number of elements (i) considered for the *CD* calculation.

### 175 **2.4.4 Potential ecological risk (PER)**

176 The potential ecological risk (*PER*) is a factor by which it is possible to assess the toxicity  
177 associated with the presence of heavy metals (Kerolli-Mustafa et al., 2015; Liu et al., 2021), that  
178 is, Cd, Cr, Cu, Ni, Pb and Zn, as well as As (metalloid) (a total of seven elements). The *PER* for  
179 an element “i” is calculated by the following equation:

$$PER_i = CF_i \cdot T_i \quad (4)$$

180

181 where  $CF_i$  is the contamination factor of the element (i), and  $T_i$  is the toxicity factor for this  
182 element, that is,  $T_i$  is based on the principle of abundance, which indicates that the potential  
183 toxicological effect of an element is proportional to its abundance, or rarity, in nature. The  $T_i$  has  
184 a specific value for each heavy metal and As (Kerolli-Mustafa et al., 2015; Liu et al., 2021).

### 185 **2.4.5 Potential toxicity of a sample (PT)**

186 Regarding the potential risk that a sample or material can produce, potential toxicity index (*PT*)  
187 is defined, and it is calculated as follows (Kerolli-Mustafa et al., 2015):

$$PT = \sum_i PER_i \quad (5)$$

## 188 **2.5 Data treatment**

189 In the case of the data treatment, a principal component analysis (PCA) was performed providing  
190 the data interpretation by using the XLSTAT software (Lumivero, 2023). Firstly, the calculation  
191 of the correlation matrix of the variables was carried out, starting from the matrix of original data.  
192 Then, the calculation of the factor matrix, starting from the Pearson’s “r” of the previous matrix.

193 **3. Results and discussion**

194 For the cores taken from the Piedras River and Bacuta Island, firstly a granulometric analysis was  
195 done to evaluate the pollutants sorption capacity of the samples, and then, concentrations of stable  
196 elements (major and trace elements) were measured. Afterwards, the suitability of the Piedras  
197 River as a sedimentary background was tested, and the environmental impact in Bacuta Island  
198 was assessed. Finally, correlation analysis was carried out for Piedras River and Bacuta Island in  
199 order to find the influence of the pollution sources on both sedimentary systems.

200 **3.1 Baseline for estuaries from the Southwest of Spain**

201 **3.1.1 Granulometry**

202 The granulometric analysis is very useful to have information on the predominant grain size for  
203 each sedimentary systems and, therefore, information on the predominant specific surface area of  
204 the grains presents in the samples, which favors the sorption of pollutants as the grain size  
205 decreases.

206 The granulometry distribution is very similar for most of the core samples, and an example of this  
207 distribution in Fig. A.2a (Supplementary Material – Appendix A) is shown (depth = 16 cm,  
208 Piedras River). Thus, it is possible to find a maximum at about 10  $\mu\text{m}$  which suggests that a fine  
209 fraction is predominant, being the fine fraction (clay + silt) about 80% of the samples. However,  
210 it is also possible to find other two small relative maximums at about 300  $\mu\text{m}$  and 500  $\mu\text{m}$ , which  
211 suggests that coarse fractions are also present, especially in the deepest samples (52 cm), with  
212 about 40% of sand.

213 Then, in Fig. A.2b it is possible to observe the percentages of the different granulometric fractions  
214 obtained for the five selected samples of the core (depths = 2 cm, 16 cm, 30 cm and 52 cm)  
215 analyzed by laser diffraction in the case of the core taken from the Piedras River. As can be  
216 observed in Fig. A.2b, the silt fraction ranged from 54% (depth = 52 cm) to 88% (depth = 16 cm).  
217 However, it is also possible to observe a significant presence of the different types of sand, which  
218 is especially true for fine and medium sands that ranged from 3% (16 cm) to 29% (52 cm), and  
219 from 1% (16 cm) to 14% (52 cm), respectively. In addition, for 2 cm of depth, it is also possible  
220 to observe a significant contribution of coarse sand which was found to be about 10%.

221 Consequently, according to the results shown in Fig. A.2, it is possible to note that the presence  
222 of the coarse fraction is relative significant for all the samples excepting for depth = 16 cm.

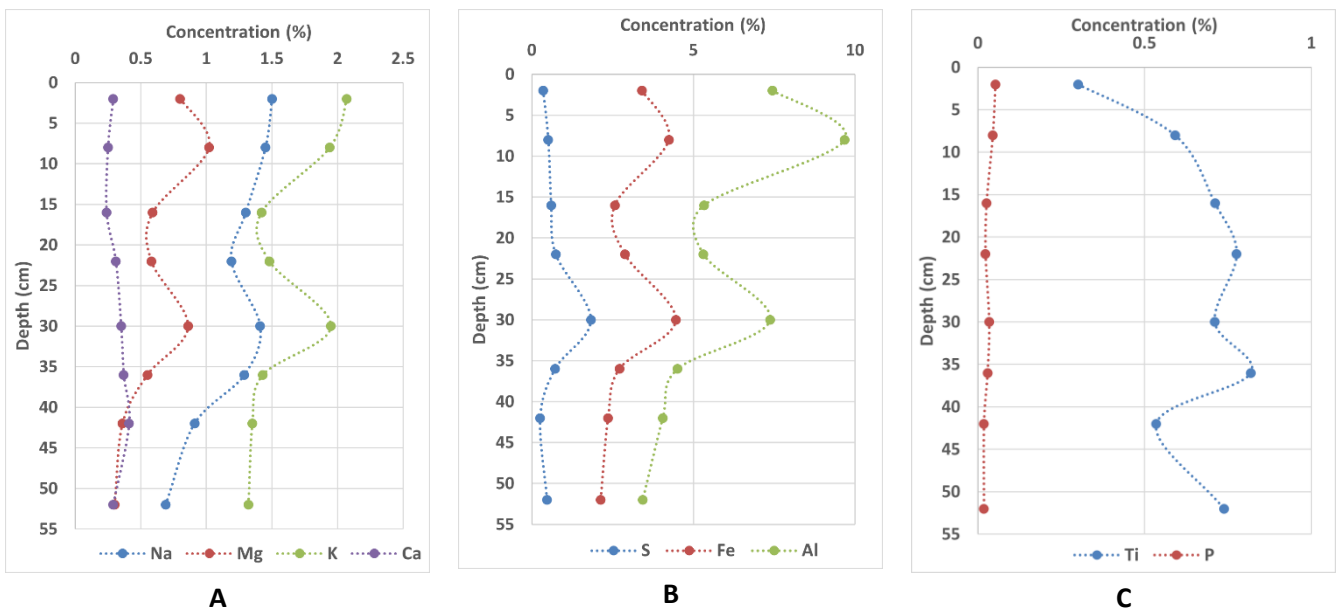
223 **3.1.2 Concentrations of major and trace elements**

224 The concentrations of the major and trace elements were obtained for the different depths  
225 analyzed for the core taken from the Piedras River. As can be seen in Fig. 2, the concentrations  
226 obtained for the major elements are similar to those found in typical soils (Rudnick and Gao,  
227 2003), where the concentrations of K, Na and Mg (Fig. 2a) were ranged from 1% to 2%, which  
228 are concentrations for typical soils. This is consistent with the granulometric analysis (see Fig.  
229 A.2), for which coarse fraction (fine sand + medium sand + coarse sand) was found to be relatively  
230 high, that is, the  $\text{SiO}_2$  content must also be relatively high, as occurred for the Piedras River core.  
231 In the case of Fe and S (Fig. 2b), note that their concentrations ranged from 2% (52 cm) to 4.5%  
232 (30 cm), and from 0.3% (42 cm) to 1.8% (30 cm), respectively, are slightly higher than typical  
233 soils. This is consistent since the Piedras River basin contain some areas belonging to the Iberian  
234 pyrite belt (IPB), which is characterized by the presence of massive polymetallic sulphurs.

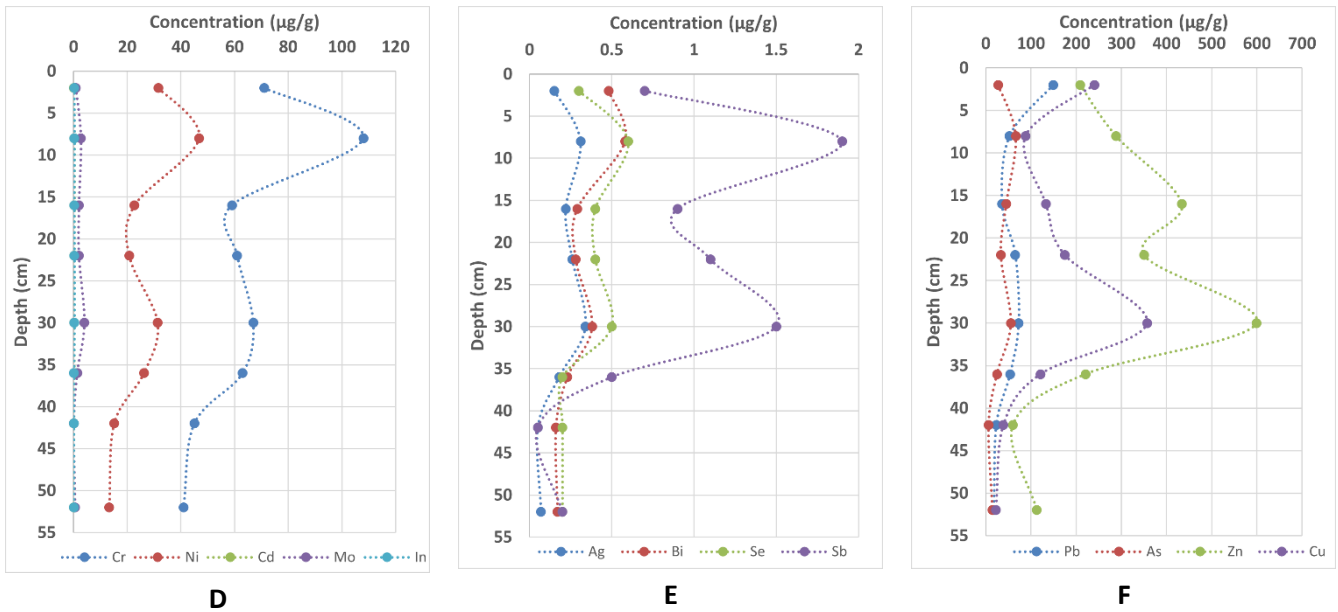
235 Furthermore, in Fig. 2b it is also possible to observe that the concentration of Al is relatively low  
 236 at the deepest layers, as 52 cm of depth. This agrees well with the granulometric fractions obtained  
 237 at 52 cm of depth, where contributions of fine and medium sands were found to be higher than  
 238 those found for a typical soil. Therefore, the percentage of SiO<sub>2</sub> present at that depth is also higher  
 239 than that found for a typical soil, making the Al concentration be lower than for a typical soil.  
 240 Then, in the case of the Ti and P (Fig. 2c), no significant concentrations were found, which were  
 241 less than 0.8% and 0.05%, respectively, for all the depths. In addition, note that for all the great  
 242 majority of the major elements, two maximums (8 cm and 30 cm of depth) were observed which  
 243 must correspond to a natural enrichment.

244 In the case of the trace elements (Figs. 2d, 2e and 2f), their concentrations were in general at the  
 245 same order of magnitude that to the unperturbed sediments, which agrees well with the results  
 246 obtained with major elements and granulometric fractions. There were only two exceptions, Cu  
 247 and Zn, whose concentrations were ranged from 100 µg g<sup>-1</sup> to 350 µg g<sup>-1</sup>, and 100 µg g<sup>-1</sup> to 600  
 248 µg g<sup>-1</sup>, respectively. However, this is consistent with geographical location of the Piedras River,  
 249 which is located near the IPB where massive polymetallic sulphurs are present. In addition, note  
 250 that for the great majority of the trace elements, two maximums were also found for the same  
 251 depths than those observed for major elements, which is consistent.

252





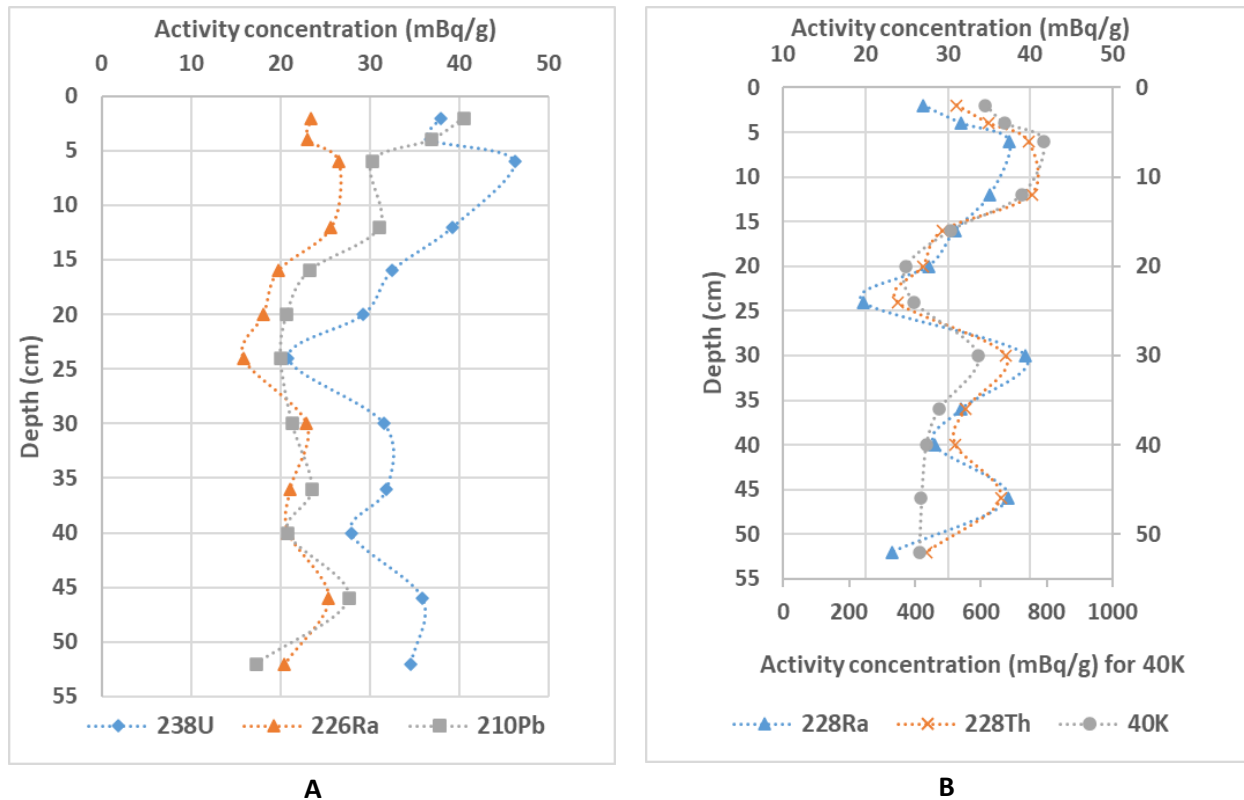


**Fig. 2.** Concentrations of both major (from A to C) and trace (from D to F) elements contained in the core taken from Piedras estuary.

253

254 **3.1.3 Concentrations of natural radionuclides**

255 In the case of the concentrations obtained for natural radionuclides ( $^{238}\text{U}$ -series,  $^{232}\text{Th}$ -series and  
 256  $^{40}\text{K}$ ) at the different depths considered for the core taken from the Piedras River, they were shown  
 257 in Fig. 3. As can be seen in Fig. 3a, the concentrations of the different radionuclides belonging to  
 258 the  $^{238}\text{U}$ -series ( $^{238}\text{U}$ ,  $^{226}\text{Ra}$  and  $^{210}\text{Pb}$ ) were ranged from  $21 \text{ Bq kg}^{-1}$  to  $46 \text{ Bq kg}^{-1}$ , from  $16 \text{ Bq kg}^{-1}$   
 259  $^1$  to  $27 \text{ Bq kg}^{-1}$ , and from  $17 \text{ Bq kg}^{-1}$  to  $41 \text{ Bq kg}^{-1}$ , respectively. These concentrations agree well  
 260 with those found for typical soils (UNSCEAR, 2000). For  $^{232}\text{Th}$ -series ( $^{228}\text{Ra}$  and  $^{228}\text{Th}$ ) and  $^{40}\text{K}$   
 261 (Fig. 3b), activity concentrations ( $20\text{-}37 \text{ Bq kg}^{-1}$ ,  $24\text{-}40 \text{ Bq kg}^{-1}$  and  $373\text{-}789 \text{ Bq kg}^{-1}$  for  $^{228}\text{Ra}$ ,  
 262  $^{228}\text{Th}$  and  $^{40}\text{K}$ , respectively) were also found to be very similar to those present for typical soils.  
 263 This is very consistent with the concentrations obtained for stable elements (see Fig. 2) and  
 264 granulometric fractions (see Fig. A.2). Furthermore, in Fig. 3, it is also possible to observe  
 265 maximums of concentrations at about 8 cm and 30 cm, which agree well with those maximums  
 266 found for stable elements.



**Fig. 3.** Concentrations of radionuclides, belonging to the  $^{238}\text{U}$ -series (A), and to the  $^{232}\text{Th}$ -series and  $^{40}\text{K}$  (B), present in the core taken from Piedras estuary.

267

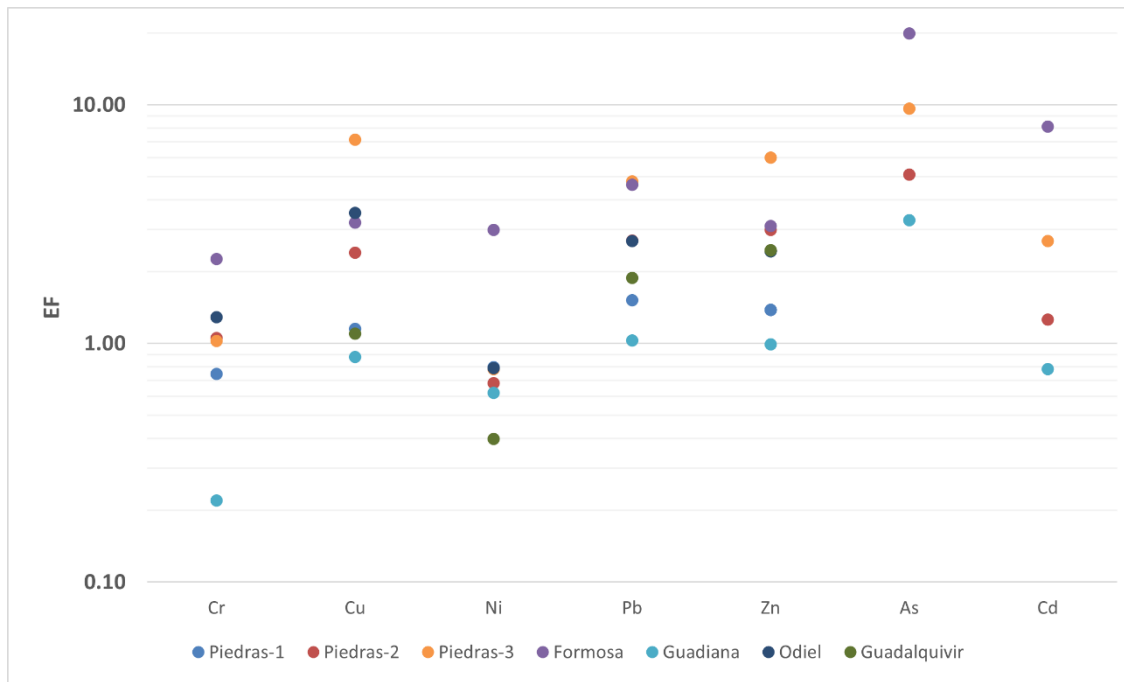
### 268 3.1.4 Background assessment

269 To decide if the Piedras River can be considered as a proper sedimentary background (baseline),  
 270 it is necessary to assess different types of pollution indexes (such as *EF* and *CF*) for stable  
 271 elements and radionuclides, allowing us to make that decision. For this, it is necessary to use of  
 272 the concentrations of the stable elements (heavy metals in this study) and radionuclides.

273 In the case of the enrichment factor, *EF*, it was calculated for the heavy metals and As. The *EF*  
 274 obtained in our case were compared with those resulting from other sedimentary systems present  
 275 in geographical regions close to the study area, being selected the estuaries of Formosa, Guadiana,  
 276 Odiel and Guadalquivir for comparison with our reference area (Piedras estuary). Piedras River  
 277 has been selected since previous works (Caliani et al., 1997; Lario et al., 2016) have demonstrated  
 278 that no mining activities has been developed on its basin. In addition, for these sedimentary  
 279 systems the data given by Rudnick and Gao (2003) for the mean of the upper earth crust was  
 280 chosen as reference, and Al was considered as the normalizer element because of its conservative  
 281 behaviour (Lee et al., 2021; Zhang et al., 2016). The *EF* for heavy metals and As were shown for  
 282 the different sedimentary systems in Fig. 4. In our case, we distinguished two cases for our studied  
 283 core, that is, Piedras-2 (two deepest samples) and Piedras-3 (entire core).

284 As can be seen in Fig. 4, *EF* was calculated in relation to Rudnick and Gao (2003) for different  
 285 estuary systems. The *EF* obtained for great majority of the stable elements were less than 2. This  
 286 result demonstrates that there is no significant pollution for any of them, where similar *EF* values

287 were found for all the cases. This is especially true when comparing our case (Piedras-2 and  
 288 Piedras-3) with Piedras-1 (Lario et al., 2016) since the core Piedras-1 was taken from a region  
 289 very close to that corresponding to the core selected for our study. Furthermore, note that in the  
 290 case of Piedras-2, lower *EF* were achieved comparing with those resulting from Piedras-3.  
 291 Therefore, Piedras River is a proper background, where Piedras-2 (average between the two  
 292 deepest samples) is a more appropriate option, but the concentrations of all elements is not known,  
 293 and for that we have taken the element concentrations of our study as baseline (background) of  
 294 the estuaries located at the Southwestern Spain.



**Fig. 4.** Enrichment factors (*EF*) calculated in relation to Rudnick and Gao (2003) for different estuary systems. Piedras-1: Deep core (several meters) of Piedras estuary (Lario et al., 2016); Piedras-2: Core of Piedras estuary, considering the two deepest samples (this study); Piedras-3: Core of Piedras estuary, considering the entire core (this study); Formosa: Surface samples from Formosa estuary (Sousa et al., 2019); Guadiana: Surface samples from Guadiana estuary (Delgado et al., 2012); Odiel: Surface samples from Odiel estuary (Borrego et al., 2002); Guadalquivir: Surface samples from Guadalquivir estuary (Riba et al., 2002).

295 In the case of the natural radionuclides ( $^{238}\text{U}$ ,  $^{228}\text{Ra}$ ,  $^{228}\text{Th}$  and  $^{40}\text{K}$ ), *EF* and *CF* were calculated  
 296 for the entire core, 12 samples (see Fig. A.3 in Supplementary Material – Appendix A). As can be  
 297 seen in Fig. A.3a, the *EF* values were generally between 1 and 2, therefore, according to Table  
 298 A.1, no significant pollution due to radioactivity was found. For further corroboration, *CF* was  
 299 also obtained for the same radionuclides and depths than the previous ones, achieving *CF* values  
 300 of 1.5 for the great majority of the cases (Fig. A.3b). Consequently, it is possible to conclude that  
 301 the Piedras River is a proper background in our case for natural radionuclides in estuarine  
 302 sediments from this geographical area.

303 In addition, to establish a proper baseline, numerical information of the concentrations of all the  
 304 stable elements and radionuclides measured was included. In Table A.2 (see Supplementary  
 305 Material – Appendix A), concentrations of major elements and radionuclides for all the depths

306 considered for this core can be consulted, as well as their average values and their respective  
 307 standard deviations (SD) and standard deviations of the average (SU). Analogously to Table A.2,  
 308 Table A.3 (see Supplementary Material – Appendix A) shows numerical information about the  
 309 concentrations of trace elements for all the considered depths, as well as their average values.  
 310 Thus, Table 1 shows the concentrations of major elements, natural radionuclides and trace  
 311 elements established as background values for the Piedras estuary core analyzed in this study. For  
 312 this, the average between the two deepest samples of the Piedras estuary core was considered (that  
 313 is, Piedras-2 in Fig. 4) since it is a more appropriate option as sedimentary background, as  
 314 previously demonstrated. In addition, the sedimentation rate of the Piedras core analyzed in this  
 315 study was obtained by the  $^{210}\text{Pb}$  dating method (San Miguel et al., 2001),, obtaining a  
 316 sedimentation rate about  $0.08 \text{ cm yr}^{-1}$ , which shows that the age of its two deepest layers (42 cm  
 317 and 52 cm) is about 525 and 650 years, respectively. Consequently, since for those sediment ages  
 318 there was no pollution source, this further corroborates that the establishment of the sedimentary  
 319 background by using the Piedras River core analyzed in this study has been properly carried out.

**Table 1**

Concentrations of major elements (in %), natural radionuclides (in  $\text{Bq kg}^{-1}$ ) and trace elements (in  $\mu\text{g g}^{-1}$ ) established as background values for the Piedras estuary core analyzed in this study.

Na	Mg	Al	K	Ca	Fe	Ti	P	S	$^{226}\text{Ra}$	$^{232}\text{Th}$	$^{40}\text{K}$	
0.80	0.33	3.74	1.34	0.35	2.25	0.64	0.018	0.37	20.8	27.4	442	
Cd	Cr	Ni	Ag	Bi	Se	Zn	As	Mo	In	Sb	Cu	Pb
< 0.1	43	14	0.06	0.17	0.20	87	11	0.36	< 0.1	0.13	21	31

320 At a national level, the Spanish Royal Decree 9/2005 establishes the guideline levels for  
 321 contaminated soils that require remediation measurements are those with heavy metal  
 322 concentrations over 100-times baseline values (Royal Decree, 2005). No studies have been carried  
 323 out to assess the base line of the estuaries located at the Southwestern Spain for major elements,  
 324 trace elements, metalloids and rare earth elements. Therefore, in Section 3.1.5 additional analysis  
 325 are carried out to further verify that the Piedras River core used in this work is not impacted.

### 326 **3.1.5 Correlation analysis and pollution sources**

327 To find possible contributions of the different potential pollution sources existing in this  
 328 geographical area (AMD, industries, wastewaters releases, etc.), a correlation analysis was  
 329 accomplished between the different stable elements. The criteria followed for the selection of  
 330 these elements was a obtain a significant correlation coefficient at level of 0.05, and R higher than  
 331  $|0.8|$ . The spearman correlation matrix in Supplementary Material – Appendix B has been  
 332 included, and the plots of the most relevant correlations in Fig. A.4 (see Supplementary Material  
 333 – Appendix A) have been included.

334 As can be seen in Fig. A.4, the correlation between Mg and Fe is positive. In the case of the  
 335 Piedras River, the Mg is a naturally occurring element. This suggests that for the Piedras River,  
 336 an enrichment of Fe is also naturally occurring, which is consistent given that Piedras River is not  
 337 significantly affected by pollution sources as proven in Sections from 3.1.2 to 3.1.4, existing only  
 338 natural enrichments due to the mixing of seawaters with the fluvial ones. In the case of the Th and  
 339 Fe, analogous behavior to the previous one was found, that is, a positive correlation. Therefore,  
 340 this suggests that Th is naturally enriched since Fe was proven not to be contributed by no

341 pollution source in the case of the Piedras River. Then, for the As-Fe pair, it also possible to  
342 observe a positive relationship, which is consistent given that the Piedras River is located near the  
343 Iberian pyrite belt. Therefore, the As is naturally present due to the massive polymetallic sulphurs.  
344 Note that Fe was selected for these three cases because of being the tracer of the pollution coming  
345 from AMD. Consequently, no impact generated by AMD was found in the case of the Piedras  
346 River.

347 Then, as it has been demonstrated in previous studies (Bolívar et al., 2009), the P element is a  
348 very good marker of the pollution coming from fertilizers industry releases. Thus, in Fig. A.4, U,  
349 Bi and Pb were plotted versus P, achieving positive correlations for the three cases. This means  
350 that U, Bi and Pb are naturally enriched since Piedras River is not affected by PGL.

351 Then, it is interesting to highlight that the correlation between Fe and P was also relatively high  
352 ( $R = 0.71$ , see Supplementary Material – Appendix B), which can be explained due to both  
353 elements are naturally enriched, therefore both are similar origin in the case of the Piedras River.

### 354 **3.2 Odiel estuary**

355 Once the background has been established and its suitability was proven, it is possible to assess  
356 the environmental impact in the Odiel River channel. For this, an order analogous to Section 3.1  
357 was followed when applying the different analysis techniques.

#### 358 **3.2.1 Granulometry**

359 As can be seen in Fig. A.5a (see Supplementary Material – Appendix A), an example of  
360 granulometric curve (depth = 42 cm) was shown for the core taken from the Bacuta Island. Thus,  
361 a maximum for a grain size slightly higher than 10  $\mu\text{m}$  is found, which proves that fine fraction  
362 (silt in this case) is predominant. Another maximum at about 1000  $\mu\text{m}$  is also possible to observe,  
363 which is related to coarse fraction (coarse sand in this case), but its contribution can be neglected  
364 comparing with that found for fine fraction.

365 In the case of Fig. A.5b, two fractions were clearly predominant: silt and fine sand. For silt, the  
366 granulometric fractions ranged from 61% (depth = 2 cm) to 84% (depth = 42 cm) and for fine  
367 sand and, they were between 8% (depth = 42 cm) and 22% (depth = 2 cm). For all the other  
368 granulometric fractions, their respective percentages were less than 5% for the great majority of  
369 the cases. Considering the relative high percentage of fine fraction at 42 cm of depth (almost 90%  
370 considering silt + clay), it suggests that it is very likely to find a maximum of concentrations of  
371 the different pollutants (trace elements and radionuclides) at about that depth.

#### 372 **3.2.2 Concentrations of major and trace elements**

373 As can be seen in Fig. 5, the concentrations of major elements (Figs. 5a, 5b and 5c) and trace  
374 elements (Figs. 5d, 5e and 5f) were shown for the different depths considered in the case of the  
375 core taken from Bacuta Island, having selected the same stable elements than those analyzed for  
376 the Piedras River (see Fig. 2 in Section 3.1.2). In addition, the concentrations of major elements  
377 and trace elements can be consulted in Tables A.4 and A.5, respectively (Supplementary Material  
378 – Appendix A).

379 For Na, Mg, K and Ca (Fig. 5a), they have similar behaviors than those found for Piedras River  
380 (see Fig. 2a), that is, they are relatively stable along all the sediment core, and all these elements  
381 have concentrations very similar for both sedimentary systems along all vertical profiles. This is  
382 consistent since these are naturally present in both sedimentary systems. For Na, it is possible to

383 observe a slight increase at superficial layers, which could be due to the contribution of seawater  
384 contained in the pores of the sediments. Then, in the case of Fe and S (Fig. 5b), they have  
385 concentrations relatively higher than those found for Piedras River (see Fig. 2b), that is, about 2-  
386 7 times and 2-6 times, respectively. These relatively high concentrations can be due to the  
387 significant affection of AMD in Bacuta Island, where Fe can be used as a tracer of AMD influence.  
388 In the case of Fig. 5c, very similar behaviors and concentrations were obtained for Ti when  
389 comparing with the core taken from Piedras River (see Fig. 2c). This suggests that Ti is naturally  
390 present in both sedimentary systems.

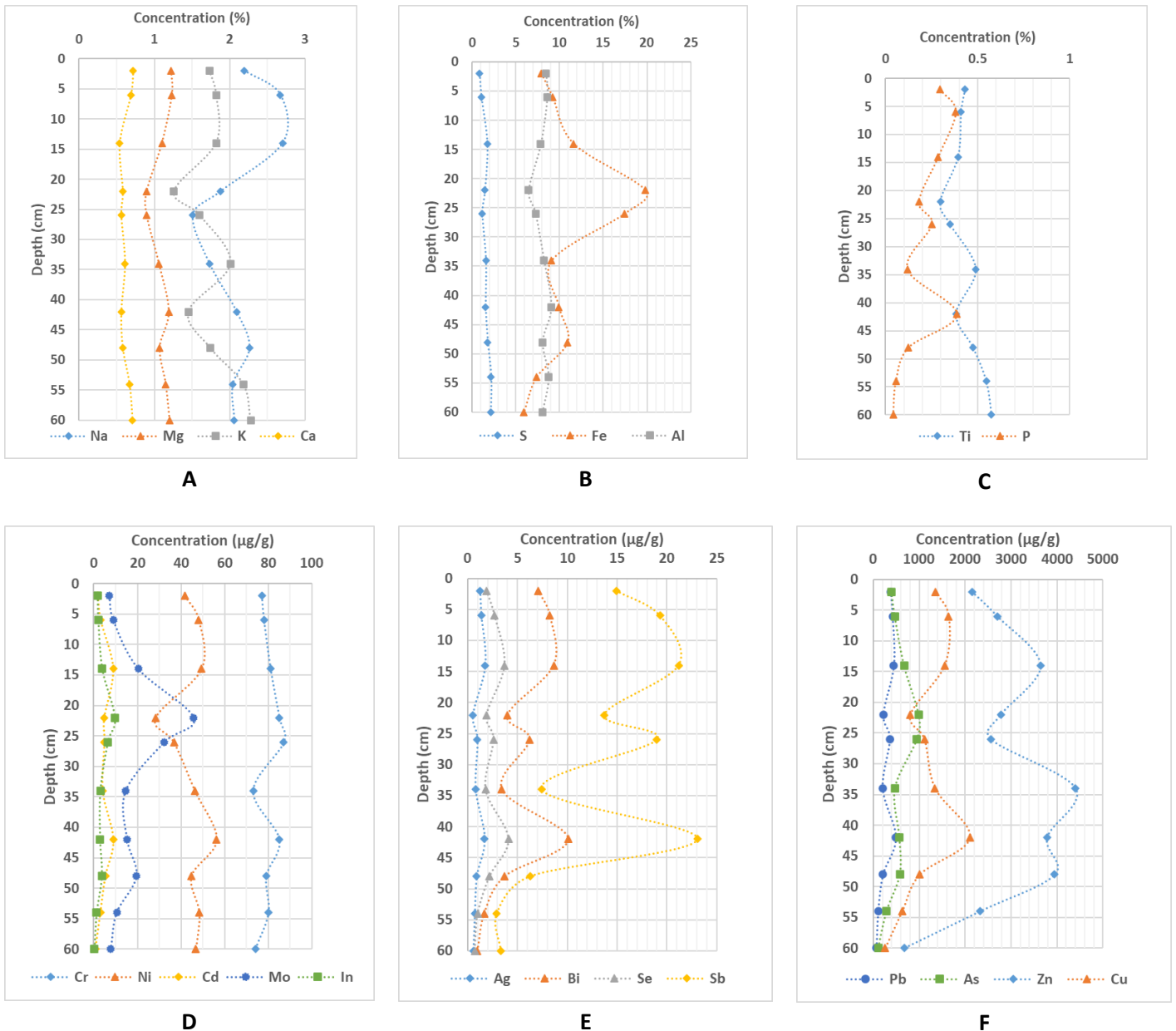
391 However, in the case of P, note that its concentrations are about one order of magnitude higher  
392 than those obtained for Piedras River, excepting the three deepest samples. This is due to the  
393 influence of the industrial production of phosphoric acid that began around the year 1965 in the  
394 surroundings of Huelva city. In this case, the pollution is caused indirectly since the P is dissolved  
395 in the phosphogypsum leachates (PGL), since the PG was pumped into the Odiel Channel in  
396 suspension of seawater (20% of PG + 80% of seawater), being P mainly in dissolution due to its  
397 very high solubility in acid water ( $\text{pH} = 1-2$ ) (Guerrero et al., 2019). Thus, P can be used as a very  
398 good tracer/marker of the pollution coming from the phosphogypsum piles and the releases of the  
399 phosphoric acid plants. In Fig. 5c, a significant increase of the P concentration up to 45 cm is  
400 observed, demonstrating this fact the impact of the phosphoric acid plants and PG leachates in  
401 this area and being possible to relate the maximum of P concentration at about 44 cm with the  
402 beginning of the industrial activity of the production of phosphoric acid, that is, about the year  
403 1965.

404 Then, in the case of the trace elements (Figs. 5d, 5e and 5f), the same stable elements than those  
405 studied in Section 3.1.2 were analyzed. For Cr, Ni, Cd, Mo and In (Fig. 5d), similar concentrations  
406 were obtained with respect to Piedras River (see Fig. 2d). However, note that in the case of the  
407 Bacuta Island core, maximum concentrations for these elements were generally obtained at 14  
408 cm, 22 cm and 42 cm of depth. This agrees well with the granulometric results (see Fig. A.5b),  
409 where the pollution level seems to be increased as the percentage of fine fraction (clay + silt)  
410 increases, finding maximums of fine fraction at 16 cm, 26 cm and 42 cm.

411 For Ag, Bi, Se and Sb (Fig. 5e), concentrations were about one order of magnitude higher than  
412 those found in Piedras River (see Fig. 2e). In addition, note that their behaviors are very similar  
413 from each other, and they are also similar to that obtained for P. This suggests that these four  
414 elements can be provided by PGL.

415 Then, in the case of the Pb, As, Zn and Cu (Fig. 5f), their concentrations are about one order of  
416 magnitude higher than those obtained for Piedras River (see Fig. 2f). For Pb, it is possible to  
417 observe a similar behavior than P, finding three maximums (14 cm, 26 cm and 42 cm). This means  
418 that Pb can come from releases of phosphogypsum piles. For the rest of elements, As, Zn and Cu,  
419 the relative high concentrations can be partially due to the AMD influence, which is consistent  
420 since they are toxic elements usually present in massive polymetallic sulphurs. In this case, this  
421 is especially true for As which follows a behaviour very similar to Fe (Fig. 5b).

422 However, note that the concentrations of all elements decreased significantly at the highest depths  
423 ( $> 55$  cm). This is consistent since the layers at those depths may have been formed before starting  
424 the mining activity, that is, before the second half of XIX century. Therefore, the concentrations  
425 of all elements tend to be similar to those obtained for unpolluted soils (Rudnick and Gao, 2003).



**Fig. 5.** Concentrations of major (from A to C) and trace (from D to F) elements present in the core taken from Bacuta Island.

426

### 427 3.2.3 Concentrations of natural radionuclides

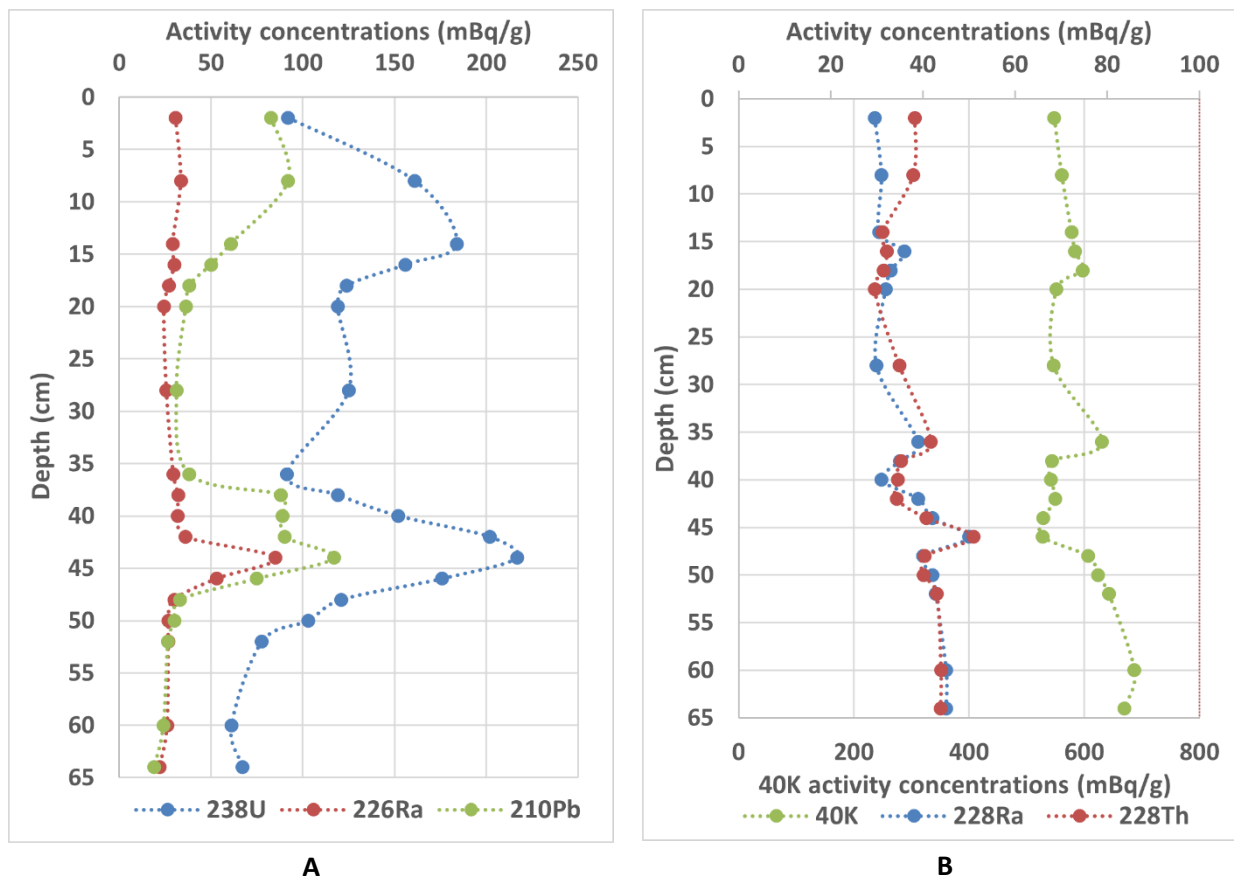
428 With respect to the behaviors and activity concentrations of the natural radionuclides along the  
 429 core taken from Bacuta Island, in Fig. 6 the concentrations of the radionuclides belonging to the  
 430  $^{238}\text{U}$ -series ( $^{238}\text{U}$ ,  $^{226}\text{Ra}$  and  $^{210}\text{Pb}$ ) (Fig. 6a), and  $^{232}\text{Th}$ -series ( $^{228}\text{Ra}$  and  $^{228}\text{Th}$ ) and  $^{40}\text{K}$  (Fig. 6b)  
 431 are shown. In addition, these concentrations of radionuclides can be consulted in Table A.6  
 432 (Supplementary Material – Appendix A). As can be seen in Fig. 6a, the  $^{238}\text{U}$  concentration pattern  
 433 is like that previously found for P (see Fig. 5c), and in the next section will be ratified by the high  
 434 correlation coefficient existing between both elements U vs P. This is consistent since U source  
 435 also comes from the phosphoric acid industry and PG piles and given that P and U are very mobile  
 436 elements, they must follow a very similar behavior/pattern. On the contrary, both  $^{226}\text{Ra}$  and  $^{210}\text{Pb}$

437 also come from phosphoric acid industry but they are very reactive, i.e., they tend to be bound to  
438 the particulate material, and they also follow a similar pattern along the core. In addition, note  
439 that the activity concentrations of  $^{238}\text{U}$ ,  $^{226}\text{Ra}$  and  $^{210}\text{Pb}$  are much higher than those obtained for  
440 the Piedras River, therefore Bacuta Island is significantly affected by PGL.

441 For  $^{232}\text{Th}$ -series ( $^{228}\text{Ra}$  and  $^{228}\text{Th}$ ) and  $^{40}\text{K}$  (Fig. 6b), it is possible to observe that they are  
442 concentrations very uniform along the core (around  $30 \text{ Bq kg}^{-1}$ ), and similar to those found for  
443 typical soils (UNSCEAR, 2000). This is consistent since in this case, no pollution sources for Th-  
444 series radionuclides are affecting to this area, therefore they are naturally present in this  
445 sedimentary system. The  $^{232}\text{Th}$  activity concentration can be calculated from the Th ICP-MS  
446 measurements and the relation  $^{232}\text{Th} (\text{Bq/kg}) = 4.05 \text{ Th} (\text{mg/kg})$ , finding an average of  $38 \text{ Bq/kg}$ ,  
447 which is very similar to the  $^{228}\text{Ra}$  one, so  $^{232}\text{Th}$ -series radionuclides are in secular equilibrium as  
448 it is expected in unperturbed estuarine sediments.

449 In the case of the Bacuta Island core, due to the clear anthropogenic impact, it was necessary to  
450 apply the methodology developed by San Miguel et al. (2001), where the  $^{230}\text{Th}/^{232}\text{Th}$  activity ratio  
451 was previously calculated and plotted versus the depth (see Fig. A.9 in Supplementary Material –  
452 Appendix A) in order to decide for which depth interval is possible to apply the  $^{210}\text{Pb}$  dating  
453 method. Thus, to make use of the  $^{210}\text{Pb}$  dating method, the  $^{230}\text{Th}/^{232}\text{Th}$  activity ratio needs to be  
454 about 1 in order to avoid possible additional contributions of  $^{210}\text{Pb}$  that can cause that the  
455 unsupported  $^{210}\text{Pb}$  cannot be properly calculated by using the equation:  $a(\text{unsupported } ^{210}\text{Pb}) =$   
456  $a(\text{supported } ^{210}\text{Pb}) - a(^{226}\text{Ra})$ , where  $a(\text{unsupported } ^{210}\text{Pb})$ ,  $a(\text{supported } ^{210}\text{Pb})$  and  $a(^{226}\text{Ra})$  are the  
457 concentrations of unsupported  $^{210}\text{Pb}$ , supported  $^{210}\text{Pb}$  and  $^{226}\text{Ra}$ , respectively. Thus, as can be seen  
458 in Fig. A.9, the  $^{230}\text{Th}/^{232}\text{Th}$  activity ratio was stable and about 1 from depths  $> 50 \text{ cm}$ . Therefore,  
459 since the concentrations of unsupported  $^{210}\text{Pb}$  are very low from depths  $> 50 \text{ cm}$ , it was not  
460 possible to apply the  $^{210}\text{Pb}$  dating method in the case of the Bacuta Island core. However, since  
461 the  $^{230}\text{Th}/^{232}\text{Th}$  activity ratio behavior was very similar to the behavior found for the vertical  
462 profile of  $^{238}\text{U}$  concentrations (Fig. 6a), it was possible to further corroborate that the depth of  
463 about  $44 \text{ cm}$  corresponds to the beginning of the industrial activity of the acid phosphoric  
464 production in Huelva, that is, the year 1965. Consequently, for the polluted part of the Bacuta  
465 Island core, it was possible to establish a sedimentation rate of about  $0.8 \text{ cm yr}^{-1}$ .





**Fig. 6.** Concentrations of radionuclides, belonging to the  $^{238}\text{U}$ -series (A), and to the  $^{232}\text{Th}$ -series and  $^{40}\text{K}$  (B), present in the core taken from Bacuta Island.

466

### 467 3.2.4 Pollution indexes

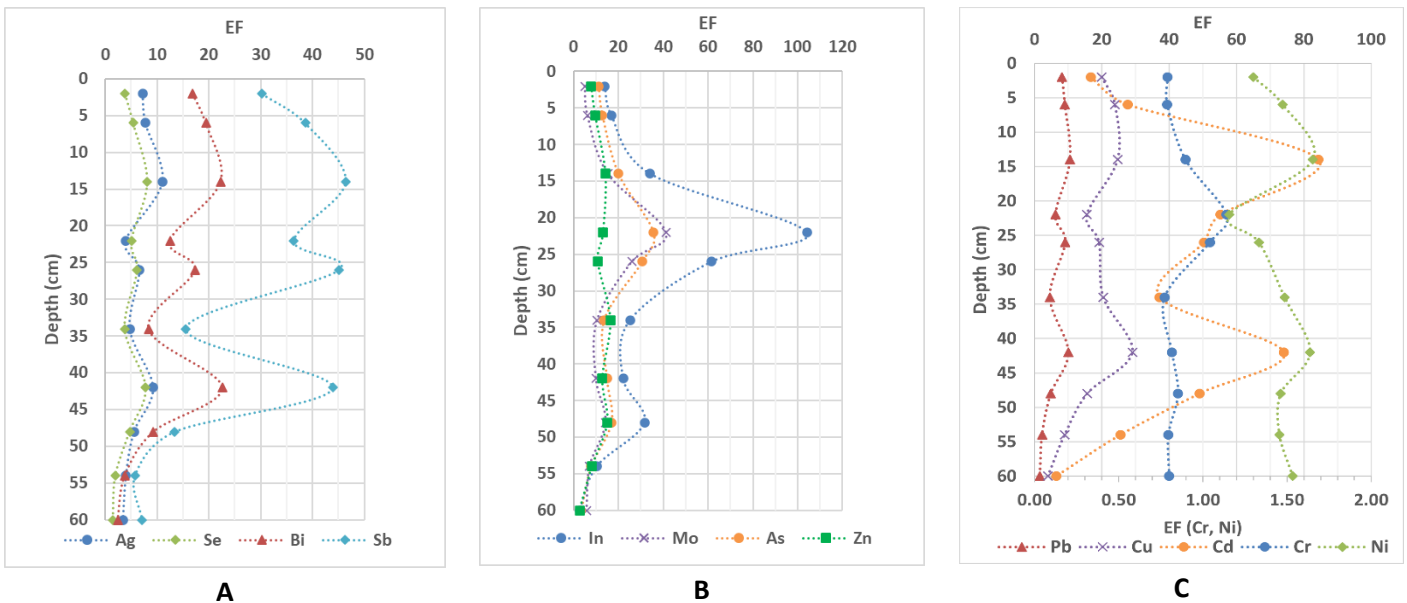
468 Once the concentrations of all stable elements and radionuclides of interest have been determined  
 469 in the case of Bacuta Island core, and the sedimentary background has been properly established  
 470 (by considering the two deepest layers of our Piedras River core, see Table 1), it is possible to  
 471 calculate the different pollution indexes in order to assess the environmental impact in the Bacuta  
 472 Island.

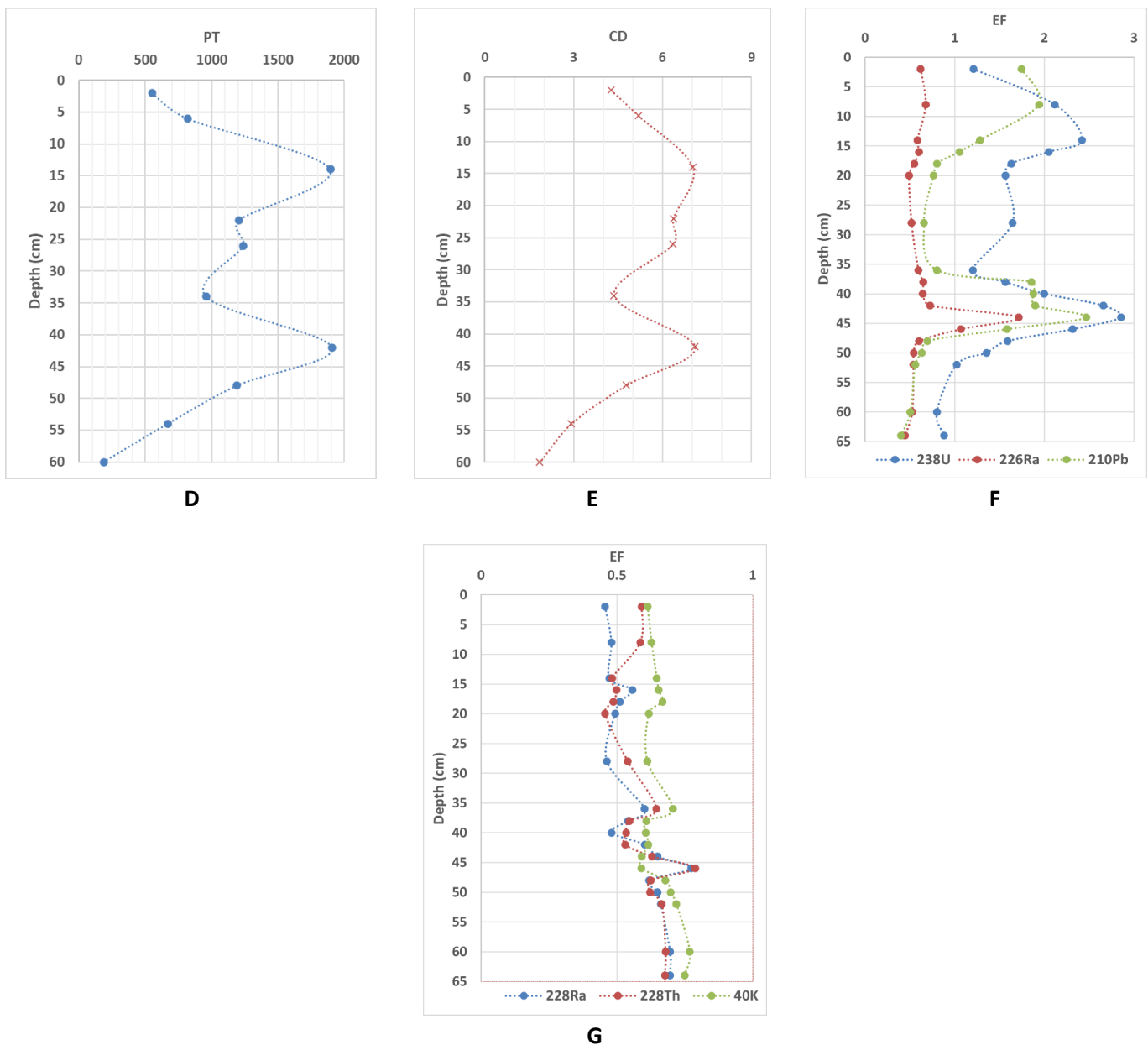
473 In the case of the *EF* for major elements, as can be seen in Figs. A.6a and A.6b (see Supplementary  
 474 Material – Appendix A), the *EF* values for Na, Mg, K, Ca and Ti were less than 2, proving that  
 475 there is no pollution associated to these major elements and they are naturally present in Bacuta  
 476 Island. In addition, note that for S, *EF* was also less than 2 along the sediment core. However, it  
 477 is necessary to clarify that it does not mean that there is no pollution related to poly-sulphide  
 478 minerals mining from the basins of the Odiel and Tinto Rivers. Due to the sulphur travels in  
 479 dissolution as sulphate anion ( $\text{SO}_4^{2-}$ ), this anion is very conservative during the mixing of the  
 480 fluvial waters and seawater, and therefore it does not precipitate during these waters mixing and  
 481 finally goes into the Atlantic Ocean, but the majority of Fe precipitates during the waters mixing  
 482 process (Hierro et al., 2013; Nieto 2013; Nieto 2007). For P and Fe (Fig. A.6c), the great majority  
 483 of *EF* values were higher than 2, which was especially true for P. This is consistent since affections  
 484 due to PGL and AMD are significantly present in Bacuta Island, where P and Fe are the main  
 485 tracers of these pollution sources, respectively.

486 Then, regarding the *CF* for the same major elements previously analyzed (Figs. A.6d, A.6e and  
 487 A.6f), it is possible to observe that results analogous to those related to *EF* were obtained, where  
 488 significant affection was mainly found for P and Fe ( $CF = 2-21$  and  $CF = 2-9$ , respectively), which  
 489 correspond to the pollution sources PGL and AMD, respectively.

490 With respect to the *EF* for trace elements (Figs. 7a, 7b and 7c), for Ag, Se, Bi and Sb (Fig. 7a),  
 491 the *EF* values were higher than 5 for all these elements at depths less than 48 cm. The fact of  
 492 obtaining relatively high *EF* values for these four elements is consistent since they are mainly  
 493 provided by PGL as previously proven (Section 3.2.2). Therefore, given that the industrial  
 494 production of phosphoric acid began in 1965, for depths higher than about 50 cm, there is no  
 495 significant contribution of this type of pollution source. In the case of the In, Mo, As and Zn (Fig.  
 496 7b), they are mainly contributed by AMD, obtaining *EF* values higher than 10 for all the elements  
 497 at any depth, excepting for surface sediments (0-5 cm of depth) and for deepest sediments (55-60  
 498 cm of depth). Then, for Pb, Cu, Cd, Cr and Ni, the *EF* values were higher than 10, excepting for  
 499 Cr and Ni, for which *EF* were close to 1 along the entire core, and excepting at the deepest layers,  
 500 where the *EF* for Pb, Cu and Cd decreased significantly.

501 In the case of the *CF* for trace elements (see Figs. A.6g, A.6h and A.6i in Supplementary Material  
 502 – Appendix A), results analogous to those found for *EF* were obtained, where all *CF* were  
 503 generally higher than 5 for all the elements, excepting for Cr and Ni, for which *CF* were very  
 504 stable and about 1.5 and 2-4, respectively, for the entire core. Furthermore, at deepest sediments,  
 505 that is, about 60 cm of depth, *CF* was generally  $\lesssim 3$ , therefore the pollution was moderate only at  
 506 about that depth.





**Fig. 7.** Enrichment factor (*EF*) (from A to C), potential toxicity (*PT*) (D) and contamination degree (*CD*) (E) obtained for the trace elements in the case of the core taken from Bacuta Island, where *EF* calculated in relation to the two deepest samples of the core taken from Piedras estuary (see Table 1), considering the Al as the normalizer element in the case of the *EF* calculations. The *EF* was also obtained for radionuclides (F and G).

507 In the case of the elements of interest from a toxicological point of view, that is, the heavy metals  
 508 and As, the index “potential ecological risk”, *PER*, was assessed for each element along the core.  
 509 As can be seen in Fig. A.7 (see Supplementary Material – Appendix A), all the *PER* values were  
 510 less than 40 ( $PER < 40$ , that is, low pollution), excepting for As and Cd, which is especially high  
 511 for Cd, where Cd and As are mainly generated by PGL and AMD, respectively.

512 When the *PT* and *CD* indexes are calculated for the different samples of the core (Figs. 7d and  
 513 7e, respectively), it is expected they follow the same pattern that Cd since its toxicity factor ( $T_i$ )  
 514 is very much higher than those for the rest of toxic elements. Thus, when using the *PT* index (see  
 515 Table A.1), it is observed a “serious” ( $300 \leq PT < 600$ ) and “very serious” ( $PT \geq 600$ ) ecological

516 risks for all the depths excepting for the deepest one, that is, at 60 cm, where the pollution was  
517 low. Then, when using the *CD* index (see also Table A.1), “high grade of pollution” was found  
518 for all the depths, excepting for depth  $\geq 55$  cm, for which the pollution was moderate.

519 In addition, the *EF* index was also assessed for natural radionuclides in the case of the core taken  
520 from Bacuta Island (Figs. 7f and 7g), as well as the *CF* index (see Figs. A.6j and A.6k in  
521 Supplementary Material – Appendix A). For the radionuclides belonging to the  $^{238}\text{U}$ -series, the  
522 *CF* was ranged from 2 to 6 for  $^{238}\text{U}$ , while for  $^{210}\text{Pb}$ , the *CF* also reached a maximum of about 6,  
523 but lower values than for  $^{238}\text{U}$  were achieved for the great majority of depths (Fig. A.6j). This is  
524 consistent since  $^{210}\text{Pb}$  has a mobility much lower than the  $^{238}\text{U}$  one. Then, the *CF* for  $^{226}\text{Ra}$  was  
525 generally  $\leq 2$ , excepting at 44 cm, which is logical since  $^{226}\text{Ra}$  has a mobility lower than the  $^{210}\text{Pb}$   
526 one. In addition, note that the behavior of the *CF* index agrees with that resulting from the activity  
527 concentrations analyzed in Section 3.2.3 (Fig. 6) as expected, finding maximums of *CF* at 10-14  
528 cm, 28 cm and 44 cm. In the case of the  $^{232}\text{Th}$ -series ( $^{228}\text{Ra}$  and  $^{228}\text{Th}$ ) and  $^{40}\text{K}$  (Fig. A.6k), the *CF*  
529 was ranged from 1 to 1.5, that is, no pollution was found due to these radionuclides, therefore  
530 they are naturally present in this sedimentary system. Regarding the *EF* index (Figs. 7f and 7g),  
531 the *EF* was higher than 2 only for  $^{238}\text{U}$  at 14 cm and 44 cm of depth, and for  $^{210}\text{Pb}$  at 44 cm. Then,  
532 for  $^{232}\text{Th}$ -series and  $^{40}\text{K}$ , the *EF* was less than 2 for the entire core, which is consistent with the  
533 *CF* values previously obtained for these radionuclides.

### 534 3.2.5 Correlation analysis and pollution sources

535 In this Section will be analyzed the different pollution sources making usage of the correlations  
536 between the different major and trace elements. Thus, according to the criteria mentioned in  
537 Section 3.1.5, several cases of high correlations were shown in Fig. A.8 (Supplementary Material  
538 – Appendix A). In the case of the couple Mg-Fe, the correlation was negative, which is the  
539 opposite case than that found for the Piedras River (see Section 3.1.5), which can be due to  
540 different basic mechanisms. Therefore, given that the Mg is naturally present in Bacuta Island,  
541 this means that Fe is being contributing anthropically, where AMD is pollution source in this case.  
542 In the case of Th, as was shown in Sections 3.2.3 and 3.2.4, the presence of Th is of natural type.  
543 Since the relationship between Th and Fe is negative, it further corroborates that Fe comes from  
544 AMD contribution. Then, for the case As-Fe, given that Fe is the tracer of AMD, and their  
545 correlation is positive, it suggests that As comes mainly from the contribution of a pollution  
546 source, which is AMD in this case as shown in Section 3.2.2 (Fig. 5).

547 On the contrary, in the case of the P, it is possible to observe that U, Bi and Pb have positive and  
548 linear correlations, but they have no significant correlations with Fe (see correlation matrix in  
549 Supplementary Material – Appendix B). This demonstrates that these three elements come mainly  
550 from the fertilizers plants activities since P is considered as the tracer of pollution coming from  
551 them.

552 In Supplementary Material – Appendix B, it is possible to observe the Principal Component  
553 Analysis (PCA) carried out for Bacuta Island. Thus, in the biplot of the principal components F1  
554 and F2, the positive axis of F1 can be identified as the affection coming from the fertilizer industry  
555 since P can be found in this axis, while Fe can be found in the positive axis of F2, thus this axis  
556 represents the influence of AMD. Thus, in the positive axis of F1, it is possible to find elements  
557 such as U, Bi and Pb, while in the positive axis of F2, As is present. This corroborates the analysis  
558 previously carried out to find the pollution sources for these four elements.

559 Then, the correlation between Fe and P was found to be not significant (see Supplementary  
560 Material) since these two elements come from different pollution sources in the case of the Bacuta  
561 Island as previously demonstrated (AMD and fertilizer industry, respectively). This behavior is  
562 opposite to that found for the Piedras River (see Section 3.1.5), which is consistent.

#### 563 **4. Conclusions**

564 An assessment of the environmental impact existing in the estuarine biosphere reserve from the  
565 Southwestern Iberian Peninsula was carried out. For this, a suitable sedimentary background was  
566 assessed for evaluating the affection existing in this biosphere reserve generated by the fertilizer  
567 complex located in the Huelva estuary, that is, between the Tinto and Odiel Rivers.

568 The following main conclusions have been found from this study:

- 569 1. The enrichment factor (*EF*) and contamination factor (*CF*) were calculated for heavy  
570 metals and natural radionuclides in the case of the Piedras River, achieving *EF* and *CF*  
571 values very close to 1 for all the cases. This proves that the Piedras River is a proper  
572 sedimentary background in our case.
- 573 2. Regarding the core taken from both Piedras estuary and Bacuta Island, it was possible to  
574 observe that clay and silt were the predominant granulometric fractions for both  
575 ecosystems. This fact demonstrates Piedras estuary is a good reference area for Odiel  
576 estuary, finding a maximum percentage of clay + silt of about 90%.
- 577 3. P and Fe were the elements whose concentrations highlighted, observing a clear indirect  
578 affection of phosphogypsum piles by means of their leachates (PGL), and the impact of  
579 acid mine drainage (AMD). Thus, P and Fe were suitable to trace the pollution coming  
580 from PGL and AMD, respectively.
- 581 4. Then, several trace elements such as Bi, Pb, Se and Sb had concentrations quite higher  
582 (about 1 order of magnitude) than those obtained for the Piedras River. These elements  
583 followed a very similar behavior than that followed by P, therefore they were mainly  
584 contributed by PGL. For As, Cu and Zn, concentrations of about 1 order of magnitude  
585 higher than those found for the Piedras River were also obtained, where the main  
586 pollution source for these elements was AMD since their behaviors were similar to the  
587 Fe one.
- 588 5. In the case of radionuclides belonging to the  $^{238}\text{U}$ -series, a clear impact by PGL was  
589 observed, and not from AMD. For radionuclides belonging to the  $^{232}\text{Th}$ -series and  $^{40}\text{K}$ ,  
590 very stable concentrations were found along the entire core, and they were similar to those  
591 related to unpolluted soils.
- 592 6. The pollution indexes *EF* and *CF*, as well as the potential ecological risk, *PER*, reached  
593 in Bacuta Island values of serious and very serious pollution for the great majority of  
594 trace elements, as well as Fe and P, but observing that in the deepest sediments (more  
595 than 50 cm depth), the pollution was much less significant, which became moderate. This  
596 dependence of the pollution on depth was possible to be observed by using the potential  
597 toxicity, *PT*, and the contamination degree, *CD*. Then, *EF* and *CF* were used for natural  
598 radionuclides, observing low-moderate pollution for radionuclides belonging to the  $^{238}\text{U}$ -  
599 series, and no pollution in the case of the  $^{232}\text{Th}$ -series and  $^{40}\text{K}$ .

600

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